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1 Basic Problems of Chemical Reaction Engineering and ...

The kinetic and the potential energy are negligible in many chemical reaction engineering applications, so Equation 15 becomes $Q V E = Q V (U + K + E P) \approx Q V U$ (16) we know that U is a function of the enthalpy, pressure, and volume, so $Q V U = Q V (H P V) = Q V X n c_{i=1} n_i h_i P Q = X n c_{i=1} F_i h_i P Q$ (17) When this is applied for both streams, the term $P Q$ cancels with the volumetric work

Introduction to Chemical Engineering: Chemical Reaction ...

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Chemical reaction engineering - Wikipedia

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