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Kelly R. Lab results: 1
doz grains of rice = 0.3
g (Use this fact as a
conversion factor.)?

grains of rice = 1.94 g

Avogadro's Number -

the mole = the number

6.022×10^{23} Molar

Mass the mass of one

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Atomic Key
mole of any pure
substance molar mass
of an element (g/mol)
equals the atomic
mass of the element in
grams molar mass of a
compound equals the
sum of the molar
masses of the
elements making up
the compound .
example problems:
Find the molar masses
of ...

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results: 1 doz grains of
rice = _____g (Use this
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results: 1 doz grains of
rice = g (Use this fact
as a conversion factor.)

? grains of rice = 1.94

g Avogadro's Number -
the mole = the

number 6.023×10^{23}

Molar Mass the mass of
one mole of any pure
substance molar mass
of an element (g/mol)

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... equals the atomic mass of the element in grams molar mass of a compound equals the sum of the molar mass of the element making up the compound example problems ...

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Mass: The terms mole
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How many formula units are in 32.6 grams of potassium oxide?

Watch the video from GPB below. As you view the video, take notes on important concepts. Pause the videos when you need more time to write down observations. Be sure to answer the video quiz questions at the end. If you would

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