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Figure 2.4.4 The Probability Density of the 4s, 4p and 3d Electrons as a Function of Radial Distance from the Nucleus The ns orbitals of the transition metals extend further from the nucleus than those of the (n-1)d orbitals, making them more available for bonding and reactions. This is illustrated here for the fourth period. Similarly the f orbitals of

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the rare earths and the lanthanides are  
...

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protons; neutrons 2. electrons 3.  
neutrons 4. electrons 5. ionic 6. The two  
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## **Compare the mass of H<sub>2</sub> expected from the reaction of ...**

Chapter 2 The Chemistry of Life Section Review 2-1 Section Review 2-4 1.

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neutrons 4. electrons 5. ionic 6. The two  
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11) Chapter 2: Chemistry of Life (Test  
Review) Covers all or most of the topics  
of Chapter 2: The Chemistry of Life A  
solution with a pH value of 12 is a \_\_\_\_\_  
Base Atoms are composed of \_\_\_\_\_  
Protons, Neutrons, and

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Chapter 2: The Chemistry of Life Biology: McDougal Littel pages 34-64. Below you find the classroom assignments and PPT's used for Chapter 2, The Chemistry of Life. You may use this website for access to PPT's, guided notes, and make

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up assignments. The Chemistry of Life Assignments.

## **Chapter 2 The Chemistry of Life - Mrs. Nicolella's Niche**

Chapter 2: Introduction to the Chemistry of Life  
Figure 2.1 Foods such as bread, fruit, and cheese are rich sources of biological macromolecules. The

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elements carbon, hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things.

## **Chapter 2: Introduction to the Chemistry of Life ...**

Chapter 2: The Chemistry of Life. STUDY.  
PLAY. atom. basic unit of matter.

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nucleus. the center of the atom which contains the protons and neutrons; in cells, structure that contains the cell's genetic material (DNA) and controls the cell's activities. electron.

## **Chapter 2: The Chemistry of Life Post-Test Review ...**

Chapter 2: The Chemistry of

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