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Chapter 2 Properties Matter Wordwise

Precipitate. A solid that forms and separates from a liquid mixture.

Element. A substance that cannot be broken down into simpler substances.

Conductivity. The ability of a material for allowing heat to flow. Pure Substance. A classification for matter that always has the same composition. Flammability.

Properties of Matter Chapter 2 WordWise Flashcards | Quizlet

Properties of Matter Chapter 2

WordWise. Solution. A mixture that results when substances dissolve to form a homogeneous mixture.

Compound. A substance that can be broken down into two or more simpler substances. Physical Change. A change which the composition of matter stays the same. Precipitate.

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Chapter 2 Properties of Matter.
WordWise. Answer the questions by
writing the correct vocabulary term in
the blanks. Use the circled letter in each
term to find the hidden vocabulary word.
Then, write a definition for the hidden
word. Clues Vocabulary Terms.

Chapter 2 Properties of Matter WordWise

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Wordwise Answer Key temperature, but
water is a liquid. Hydrogen can fuel a
fire, and oxygen can keep a fire burning,
but water does not burn or help other
substances to burn. Chapter 2 Properties
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Chapter 2 Properties Of Matter Wordwise Answer Key

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Answer the questions by writing the correct vocabulary term in the blanks. Use the circled letter in each term to find the hidden vocabulary word. Then, write a definition for the hidden word.
Clues Vocabulary Terms A mixture that results when

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Chapter 2 - Mr. Jureks Science Classes

Section 2.1: Classifying Matter. Pure Substances. Matter that always has exactly the same composition is classified as a pure substance.

Examples: salt and sugar. **Every sample of a given substance has the same properties because a substance has a fixed, uniform composition.

Ch. 2 Learning Objectives

Chapter 2 Properties of Matter Section 2.3 Chemical Properties (pages 54–58)

This section discusses chemical properties and describes clues that may show that a chemical change has taken place.

Chapter 2 Properties of Matter Section 2.3 Chemical Properties

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Start studying Chapter 2.3 WordWise. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

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Physical Science Reading and Study Workbook Level B Chapter 213. IPLS. Chapter 2 Properties of Matter. Summary. 2.1 Classifying Matter. Every sample of a given substance has the same properties because a substance has a fixed, uniform composition.

Chapter 2 Properties of Matter

Chapter 2 Properties of Matter Section 2.1 Classifying Matter (pages 38–44)
This section explains how materials are classified as pure substances or mixtures.

Chapter 2 Properties of Matter Section 2.1 Classifying Matter

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Answer the questions by writing the correct vocabulary term in the blanks Use the circled letter in each term to find the hidden vocabulary word Then, write a definition for the hidden word Clues Vocabulary Terms 9/29/2003 12:47:01 PM

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Chapter 2

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Matter Word Wise Vocabulary ...

Interactive Reader 33 Matter SECTION 2
Name _____ Class _____ Date _____
Properties of Matter
continued IDENTIFYING SUBSTANCES
Because many physical properties stay
constant, you can use them to identify a
material. For example, all samples of
pure water are colorless liquids at room
tem-perature and atmospheric pressure.
Pure water is never a powdery green
solid.

CHAPTER 2 SECTION 2 Properties of Matter

Chapter 2 Properties of Matter Section
2.1 Classifying Matter. Name _____
_____ Class _____ D
ate _____. Chapter 2 Properties of
Matter. Section 2.1 Classifying Matter.
(pages 38–44) This section explains how
materials are classified as pure
substances or mixtures. It discusses
types of pure substances and mixtures.

Chapter 2 Properties of Matter Section 2.1 Classifying Matter

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Physical Science Reading and Study Workbook Level B Chapter 2 17 Section 2. 13 Properties of Matter Physical Chemical color, shape, mass, volume, density, etc water = 16 Properties of Matter Physical Chemical describes how a substance reacts. 698 Equilibrium theories and principles apply to a variety of phenomena in nature. 604, SE These conceptual questions challenge students to apply the section content to real-world applications. heterogeneous 16.

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