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Glossary Terms •
radioactivity: the
process by which ...

Chapter 25

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CHAPTER 25 What

You'll Learn You will

trace the history of
nuclear chemistry from
dis-covery to

application. You will

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identify types of radioactive decay and solve decay rate problems. You will describe the reactions involved in nuclear fission and fusion. You will learn about applications of nuclear reactions and the effects of radiation exposure.

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800 Chapter 25 Types
of Radiation Discuss
Explain that the nuclei
of a radioactive
element spontaneously
decompose. Nuclear
chemistry is the study
of changes in matter
that originate in atomic
nuclei. Ask, What types
of radiation exist, and
how harmful are they?
(The three most
common types of
radiation emitted by

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unstable nuclei are

25.1 Nuclear Radiation 25

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NUCLEAR RADIATION 1.

What happens to the mass number and atomic number of an atom that undergoes beta decay? 2.

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• Radioactivity is the process by which nuclei emit particles and rays as they break down. • The name of the penetrating rays emitted by a radioactive source is called radiation. • A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or particles and often becoming a new element.

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Answer. ${}_{53}^{125}\text{I} + e^{-} \rightarrow {}_{52}^{125}\text{Te}$. The

two general kinds of nuclear reactions are nuclear decay reactions and nuclear transmutation reactions. In a nuclear decay reaction, also called radioactive decay, an unstable nucleus emits radiation and is transformed into the nucleus of one or more other elements.

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Radioactivity

- Radioactivity is the process by which nuclei emit particles and rays as they break down.
- The name of

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the penetrating rays emitted by a radioactive source is called radiation. • A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

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table showing some

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Mass number and atomic number are conserved in all nuclear reactions. The mass of a 25.0 g piece of ^{238}Cm (half-life: 2.4 hr) will be reduced to 147 3.1 g after 7.2 hr.

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Radioactivity

- Radioactivity is the process by which nuclei emit particles and rays as they break down.
- The name of the penetrating rays

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emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

Pearson Education Chapter 25 Nuclear Chemistry Answer Key

Nuclear radiation can ionize atoms.15. Is the following sentence true or false? One potential

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danger of radon gas is that prolonged exposure to it can lead to lung cancer.

trueDetecting Nuclear Radiation (page

297)16. Name two devices that are used to detect nuclear radiation.a. Geiger counters b. Film badges86 Physical Science Guided Reading ...

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