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An acid is a substance that contains hydrogen and ionizes to produce hydrogen ions in aqueous solution; A base is a substance that contains a hydroxide group and dissociates to

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produce a hydroxide ion in aqueous solution.

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Acids. Bases. acids react with.... to
produce.... what did geologist find
interesting abo.... -taste sour...
-conductors of electricity... -can turn
litmus paper.... -taste bitter...

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-conductors of electricity... -feels slippery... -can....

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PLAY. Acid Disassociation Constant (K_a)

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The Ratio of the concentration of the dissociated form of an acid to the undissociated form; stronger acids have larger K_a values than the weaker acids. Acidic Solution.

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Acid-the particle formed when a base gains a hydrogen ion Neutralization Reaction-a reaction in which an acid and a base react in an aqueous solution to produce a salt and water Base - a compound that... Chemistry- 19 and 20 - Chapters 19 and 20 Study Guide...

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Acids Bases

Name _____ Date _____ Class Section _____
Strengths of Acids and Bases
In your textbook, read about strengths of acids. Circle the letter of the choice that best completes the statement or answers the question.

1. Acid A and acid B are of equal concentration and are tested with a conductivity apparatus.

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strong acid? What is the pH of a 1.5×10^{-3} M solution of NaOH, a strong base? What is the molarity of a KOH solution if 25.0 mL of it is neutralized by 31.7 mL of a 0.100M nitric acid solution? Write the steps in the complete ionization of the

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following polyprotic acids. 4. H_2CO_3 5.
 H_3BO_3 A solution has a $[\text{H}^+]$ of $1.0 \times 10^{-5}\text{M}$. 17. lg. 19 ...

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are the properties of acids and bases?
acids taste sour, will change the color of
an acid-base indicator, and can be
strong or weak electrolytes in aqueous
solution.

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Use these activities to study the vocabulary and major concepts presented in this chapter. compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base and one of its salts.

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634 Chapter 18 • Acids and Bases

Section 118.18.1 Figure 18.1

Rhododendrons flourish in rich, moist soil that is moderately acidic, whereas semper-vivum, commonly called hen and chicks, grows best in drier, slightly basic soil. Introduction to Acids and

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Bases-!).)DEADifferent models help describe the behavior of acids and bases.

Chapter 18: Acids and Bases

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Enolates and enols can form C-C bonds •

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