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20 terms from Pearson's Physical Science Chapter 16. Terms in this set (20) Heat. The transfer of thermal energy from one object to another because of a temperature difference. Temperature. The measure of how hot or cold something is compared to a reference point, related to average kinetic energy.

Physical Science-Chapter 16 Thermal Energy and Matter ...

16.1 Thermal Energy and Matter Heat flows spontaneously from hot objects to

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cold objects. • Heat is the transfer of thermal energy from one object to another because of a temperature difference. Temperature is related to the average kinetic energy of the particles in an object due to their random motions through space.

Chapter 16 Thermal Energy and Heat

Section 16.1 Thermal Energy and Matter (pages 474–478) This section defines heat and describes how work, temperature, and thermal energy are related to heat. Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained.

Chapter 16 Thermal Energy and Heat Section 16.1 Thermal ...

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a ref.... Increase in volume of material when its temperature increases.

Chapter 16 Thermal Energy And Heat Section 16.1 Matter ...

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Convection (pages 480–481) 6. The
transfer of thermal energy when
particles of a fluid move from one place
to another is called . 7.

Chapter 16 Thermal Energy and Heat Section 16.2 Heat and ...

Thermal Energy and Thermal Expansion
Recall that thermal energy is the total
potential and kinetic energy of all the
particles in an object. The thermal
energy of an object depends on its
mass, temperature, and phase (solid,
liquid, or gas). The larger the mass, the
greater the thermal energy.

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Reading Strategy (page 474)

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IPLS Section 16.2 Heat and Thermodynamics (pages 479-483) This

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section discusses three kinds of thermal energy transfer and introduces the first, second, and third laws of thermodynamics. Reading Strategy (page 479)

Chapter 16 Thermal Energy and Heat Section 16.2 Heat and ...

Chapter 16 Thermal Energy and Heat Section 16.2 Heat and Thermodynamics (pages 479–483) This section discusses three kinds of thermal energy transfer and introduces the first, second, and third laws of thermodynamics.

Chapter 16 Thermal Energy and Heat Section 16.2 Heat and ...

Thermal Conduction and Surface Area Background Information The quantity of energy transferred by heat from a body depends on a number of physical properties of the body and its surroundings. For a given substance, the rate at which thermal energy is transferred by conduction depends on temperature difference, cross-sectional

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(pages 474–478) This section defines heat and describes how work, temperature, and thermal energy are related to heat. Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained.

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